

Heavy Metals in Domestic Drinking & Effluent Waters

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The Impending Water Crisis of Tampa Bay: Waste,
Reuse & Environmental Protection
USF College of Public Health



Outline

- What are heavy metals?
- Why are we concerned with them?
- What regulations apply to them?
- What are current treatment options?
- Should we be concerned with heavy metals in effluent from the Howard F. Curren Advanced wastewater treatment plant?
- Some things to consider:
 - Are you concerned with heavy metals in your tap water?
 - Are heavy metals in grey water a potential issue?
 - What about heavy metal release during Aquifer Storage & Recovery?
- Conclusions

What are “heavy metals”?

No clear definition, but usually includes metals that have some toxic effect.

PERIODIC TABLE OF THE ELEMENTS

<http://www.ktf-split.hr/periodni/en/>

PERIODIC TABLE OF THE ELEMENTS

<http://www.ktf-split.hr/periodni/en/>

PERIOD

GROUP	1	2	13	14	15	16	17	18										
PERIOD	1	2	3	4	5	6	7	8										
1	H 1.0079 HYDROGEN							He 4.0026 HELIUM										
2	Li 6.941 LITHIUM	Be 9.0122 BERYLLIUM																
3	Na 22.990 SODIUM	Mg 24.305 MAGNESIUM	B 10.811 BORON	C 12.011 CARBON	N 14.007 NITROGEN	O 15.999 OXYGEN	F 18.998 FLUORINE	Ne 20.180 NEON										
4	K 39.098 POTASSIUM	Ca 40.078 CALCIUM	Sc 44.956 SCANDIUM	Ti 47.867 TITANIUM	V 50.942 VANADIUM	Cr 51.996 CHROMIUM	Mn 54.938 MANGANESE	Fe 55.845 IRON	Co 58.933 COBALT	Ni 58.693 NICKEL	Cu 63.546 COPPER	Zn 65.39 ZINC	Ga 69.723 GALLIUM	Ge 72.64 GERMANIUM	As 74.922 ARSENIC	Se 78.96 SELENIUM	Br 79.904 BROMINE	Kr 83.80 KRYPTON
5	Rb 85.468 RUBIDIUM	Sr 87.62 STRONTIUM	Y 88.906 YTTRIUM	Zr 91.224 ZIRCONIUM	Nb 92.906 NIOBIUM	Mo 95.94 MOLYBDENUM	Tc (98) TECHNETIUM	Ru 101.07 RUTHENIUM	Rh 102.91 RHODIUM	Pd 106.42 PALLADIUM	Ag 107.87 SILVER	Cd 112.41 CADMIUM	In 114.82 INDIUM	Sn 118.71 TIN	Sb 121.76 ANTIMONY	Te 127.60 TELLURIUM	I 126.90 IODINE	Xe 131.29 XENON
6	Cs 132.91 CAESIUM	Ba 137.33 BARIUM	La-Lu 57-71 Lanthanide	Hf 178.49 HAFNIUM	Ta 180.95 TANTALUM	W 183.84 TUNGSTEN	Re 186.21 RHENIUM	Os 190.23 OSMIUM	Ir 192.22 IRIDIUM	Pt 195.08 PLATINUM	Au 196.97 GOLD	Hg 200.59 MERCURY	Tl 204.38 THALLIUM	Pb 207.2 LEAD	Bi 208.98 BISMUTH	Po (209) POLONIUM	At (210) ASTATINE	Rn (222) RADON
7	Fr (223) FRANCIUM	Ra (226) RADIUM	Ac-Lr 89-103 Actinide	Rf (261) RUTHERFORDIUM	Db (262) DUBNIUM	Sg (266) SEABORGIUM	Bh (264) BOHRNIUM	Hs (277) HASSIUM	Mt (268) MEITNERIUM	Uun (281) UNUNUNIUM	Uuu (272) UNUNVIUM	Uub (285) UNUNBIUM	Uut (289) UNUNTRIUM	Uuq (293) UNUNQUADIUM				

RELATIVE ATOMIC MASS (1)

GROUP IUPAC

GROUP CAS

ATOMIC NUMBER

SYMBOL

ELEMENT NAME

■ Metal	■ Semimetal	■ Nonmetal
1 Alkali metal	16 Chalcogens element	
2 Alkaline earth metal	17 Halogens element	
■ Transition metals	18 Noble gas	
■ Lanthanide		
■ Actinide		

STANDARD STATE (25 °C; 101 kPa)

Ne - gas **Fe** - solid

Ga - liquid **Tc** - synthetic

LANTHANIDE

La 57 138.91 LANTHANUM	Ce 58 140.12 CERIUM	Pr 59 140.91 PRASEODYMIUM	Nd 60 144.24 NEODYMIUM	Pm 61 (145) PROMETHIUM	Sm 62 150.36 SAMARIUM	Eu 63 151.96 EUROPIUM	Gd 64 157.25 GADOLINIUM	Tb 65 158.93 TERBIUM	Dy 66 162.50 DYSPROSIUM	Ho 67 164.93 HOLMIUM	Er 68 167.26 ERBIUM	Tm 69 168.93 THULIUM	Yb 70 173.04 YTTERIUM	Lu 71 174.97 LUTETIUM
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ACTINIDE

Ac 89 (227) ACTINIUM	Th 90 232.04 THORIUM	Pa 91 231.04 PROTACTINIUM	U 92 238.03 URANIUM	Np 93 (237) NEPTUNIUM	Pu 94 (244) PLUTONIUM	Am 95 (243) AMERICIUM	Cm 96 (247) CURIUM	Bk 97 (247) BERKELIUM	Cf 98 (251) CALIFORNIUM	Es 99 (252) EINSTEINIUM	Fm 100 (257) FERMIUM	Md 101 (258) MENDELEVIUM	No 102 (259) NOBELIUM	Lr 103 (262) LAWRENCIUM
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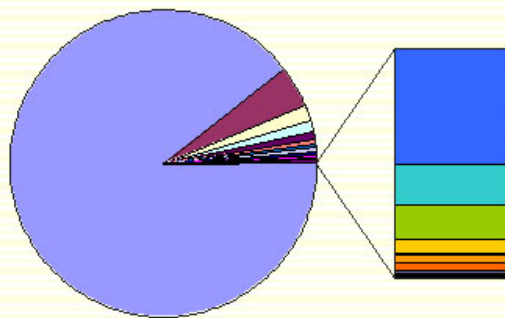
(1) Pure Appl. Chem., 73, No. 4, 667-683 (2001)
Relative atomic mass is shown with five significant figures. For elements having no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element.
However three such elements (Th, Pa, and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.

Editor: Aditya Vardhan (adivar@netlinx.com)

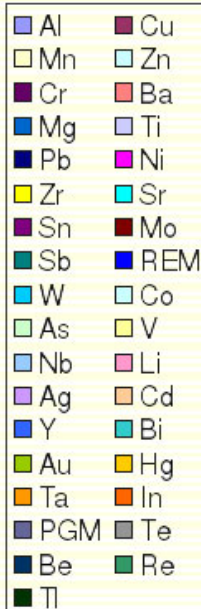
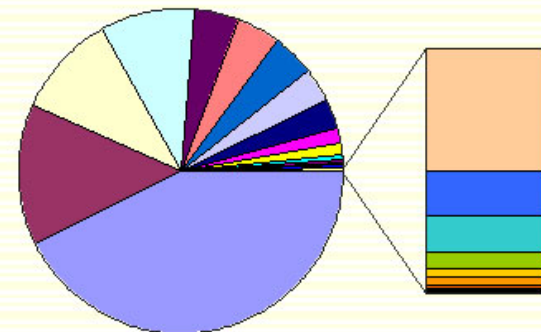
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How much heavy metals do we produce?

Annual global primary production
(more than 1 billion metric tons
of metal elements)



Annual global primary production
excluding iron (around 110 million metric tons
of metal elements)



2008 Data

Dr. A.M. Diederer, MSc., Metal minerals scarcity: A call for managed austerity and the elements of hope. Based on data from United States Geological Survey (USGS), **Mineral commodity summaries 2008**

Contaminant	Potential Health Effects from Ingestion	Sources of Contaminant
Antimony	Increase in blood cholesterol; decrease in blood sugar	Discharge from petroleum refineries; fire retardants; ceramics; electronics; solder
Arsenic	Skin damage or problems with circulatory systems, and may have increased risk of getting cancer	Erosion of natural deposits; runoff from orchards, runoff from glass & electronicsproduction wastes
Barium	Increase in blood pressure	Discharge of drilling wastes; discharge from metal refineries; erosion of natural deposits
Beryllium	Intestinal lesions	Discharge from metal refineries and coal-burning factories; discharge from electrical, aerospace, and defense industries
Cadmium	Kidney damage	Corrosion of galvanized pipes; erosion of natural deposits; discharge from metal refineries; runoff from waste batteries and paints
Chromium (total)	Allergic dermatitis	Discharge from steel and pulp mills; erosion of natural deposits
Copper	Short term exposure: Gastrointestinal distress	Corrosion of household plumbing systems; erosion of natural deposits
	Long term exposure: Liver or kidney damage	
	People with Wilson's Disease should consult their personal doctor if the amount of copper in their water exceeds the action level	
Lead	Infants and children: Delays in physical or mental development; children could show slight deficits in attention span and learning abilities	Corrosion of household plumbing systems; erosion of natural deposits
	Adults: Kidney problems; high blood pressure	
Mercury (inorganic)	Kidney damage	Erosion of natural deposits; discharge from refineries and factories; runoff from landfills and croplands
Selenium	Hair or fingernail loss; numbness in fingers or toes; circulatory problems	Discharge from petroleum refineries; erosion of natural deposits; discharge from mines
Thallium	Hair loss; changes in blood; kidney, intestine, or liver problems	Leaching from ore-processing sites; discharge from electronics, glass, and drug factories

Types of reuse applications

We reclaim water from wastewater treatment plants for various reuse applications.

- Unrestricted Urban
 - Irrigation
 - Toilet Flushing
 - Fire Protection
 - Construction
 - Landscape Impoundment
 - Street Cleaning
- Restricted Urban
 - Golf courses
- Agricultural (Food Crops)
- Agricultural (Non-food Crops)
- Unrestricted Recreational
- Restricted Recreational
- Environmental (Wetlands)
- Industrial
- Groundwater Recharge (Nonpotable Aquifer)
- Indirect Potable Reuse

various water quality criteria and regulations

EPA Recommended Limits for Heavy Metals in Reclaimed Water for Irrigation

Constituent	Long-Term Use (mg/l)	Short-Term Use (mg/l)	Up to 20 years
Aluminum	5.0	20	
Arsenic	0.10	2.0	
Cadmium	0.01	0.05	
Chromium	0.1	1.0	
Cobalt	0.05	5.0	
Copper	0.2	5.0	
Iron	5.0	20.0	
Lead	5.0	10.0	
Manganese	0.2	10.0	
Molybdenum	0.01	0.05	
Nickel	0.2	2.0	
Selenium	0.02	0.02	
Vanadium	0.1	1.0	
Zinc	2.0	10.0	

<http://epa.gov/nrmrl/pubs/625r04108/625r04108chap2.pdf>

Treatment Technologies

Selected Component to be removed	Drinking Water Standard	Coagulation, Sedimentation	Lime Softening	Adsorption			Ion Exchange		Membrane Filtration	
				GAC	PAC	Activated Alumina	Anion	Cation	Reverse Osmosis	Ultrafiltration
Arsenic (+3)	0.010	G-E	F-E	F-G	P-F	F-E	G-E	P	E	P
Arsenic (+5)	0.010	G-E	F-E	F-G	P-F	F-E	G-E	P	E	F
Copper	1.3	G	G-E	F-G	P	-	P	F-G	E	-
Lead	0.015	E	E	F-G	P-F	P	P	F-G	G-E	-
Mercury	0.022	F-G	F-G	F-G	P	P	P	F-G	G-E	-

Performance rating at removing selected component: E – Excellent, G – Good, F – Fair, P – Poor.

Removal of heavy metals after various treatment stages in wastewater treatment

	Raw Conc. µg/L	Primary Effluent		Secondary Effluent		Tertiary Effluent		Advanced Water Treatment Effluent		Overall %R
		Conc. µg/L	%R	Conc. µg/L	%R	Conc. µg/L	%R	Conc. µg/L	%R	
Arsenic	3.2	3.1	3	2.5	19	1.5	30	3	40	92
Cadmium	0.6	0.5	17	12	0	0.1	67	0.1	0	83
Chromium	3	4	0	2	32	1	24	1	25	83
Copper	63	79	0	43	33	9	52	11	0	83
Iron	600	534	11	180	59	50	22	40	2	94
Lead	5	6	0	6	0	1	93	1	0	91
Mercury	0.3	2	33	0.1	33	1	0	0.1	0	67

Water reuse: Issues, technologies and applications. Takashi, A. et al., (2007). Metcalf & Eddy.

Are metals a concern in effluent from Howard F. Curren Advanced Wastewater Treatment plant in Tampa, FL?

	Ni ug/l	Hg ug/l	Ag ug/l	Cu ug/l	Ni ug/l	Pb ug/l	Zn mg/l
Effluent 1/13/10	3.7	<0.04	<4.0	<8	5.0	<12.0	0.014
SWQS Permit Level	8.3		2.3	3.7		8.5	0.086
Freshwater chronic	52.0	0.77			52.00	2.5	0.120
SDWA MCL		2		1300		15	

	Tl ug/l	Al mg/l	As mg/l	Ba mg/l	Be mg/l	Ca mg/l	Co mg/l
Effluent 1/13/10	<52	<0.060	<0.012	0.011	<0.001	96.6	<0.004
SWQS permit Level	6.3	1.5	0.05		0.0001		
Freshwater chronic		0.087	0.15				
SDWA MCL	2		0.01	2.000	0.004		

	Cd mg/l	Cr mg/l	Fe mg/l	K mg/l	Mg mg/l	Mn mg/l	Mo mg/l
Effluent 1/13/10	<0.001	<0.008	0.049	16.3	27.40	0.010	0.004
SWQS permit Level	0.009	0.08					
Freshwater chronic	0.0003	0.011	1.000				
SDWA MCL	0.005	0.1					

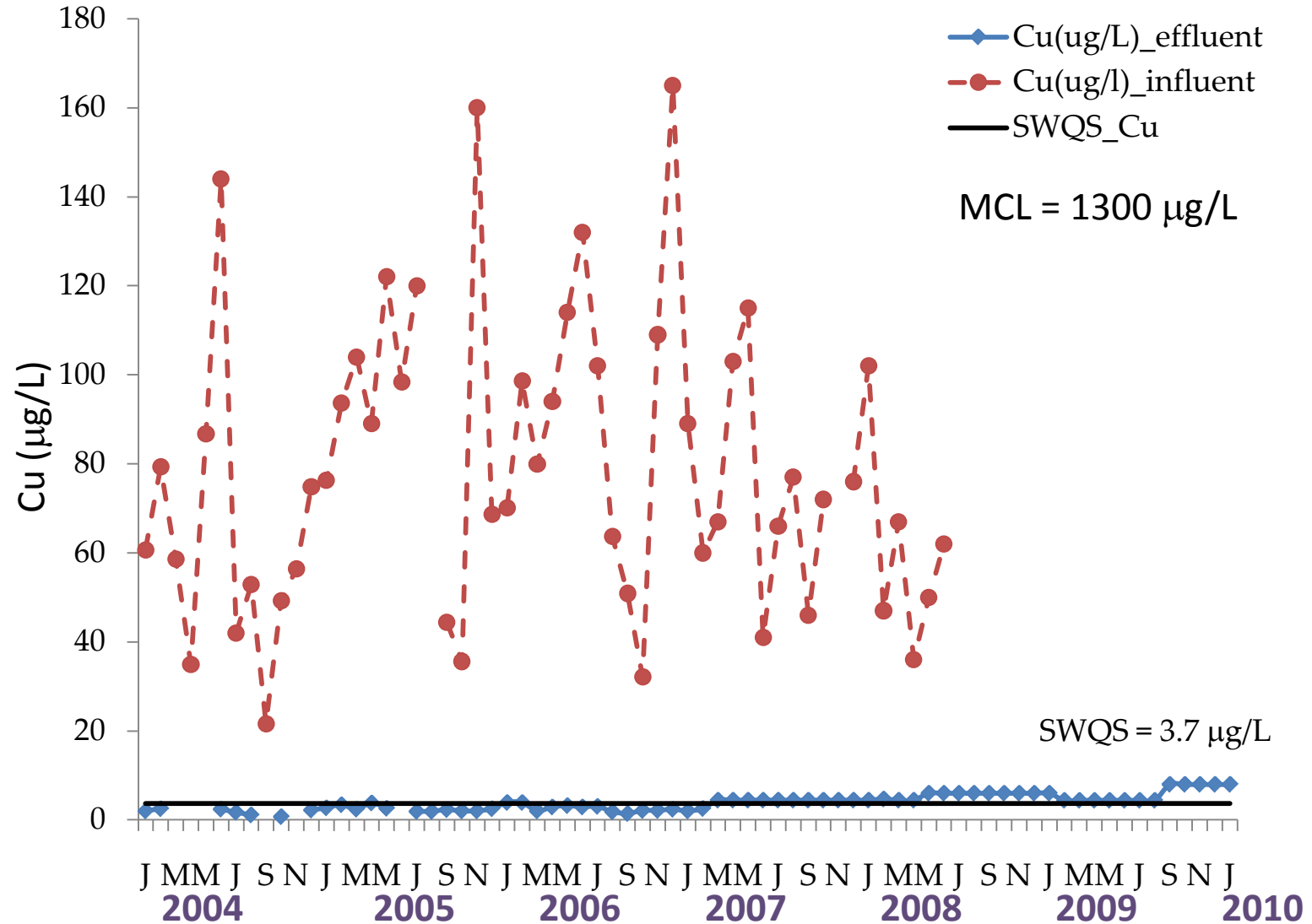
	Na mg/l	Sb mg/l	Se mg/l	Sr mg/l	V mg/l
Effluent 1/13/10	193.0	<0.024	<0.016	0.450	<0.004
SWQS permit Level		4.3	0.071		
Freshwater chronic			0.005		
SDWA MCL			0.05		

The detection limit at the lab for As is 12 µg/l.

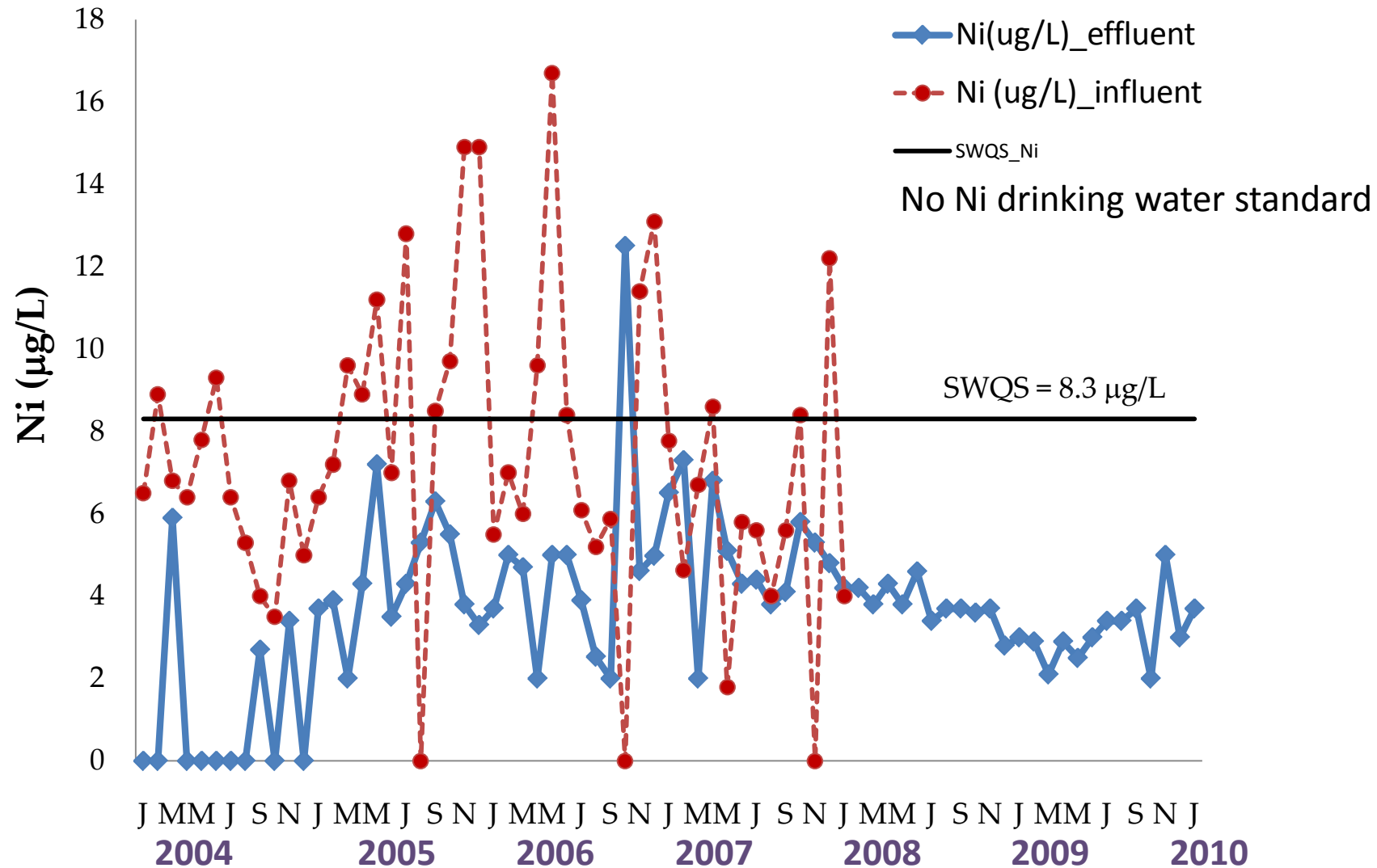
Meeting the SWQS permit level

Don't know if meets SDWA

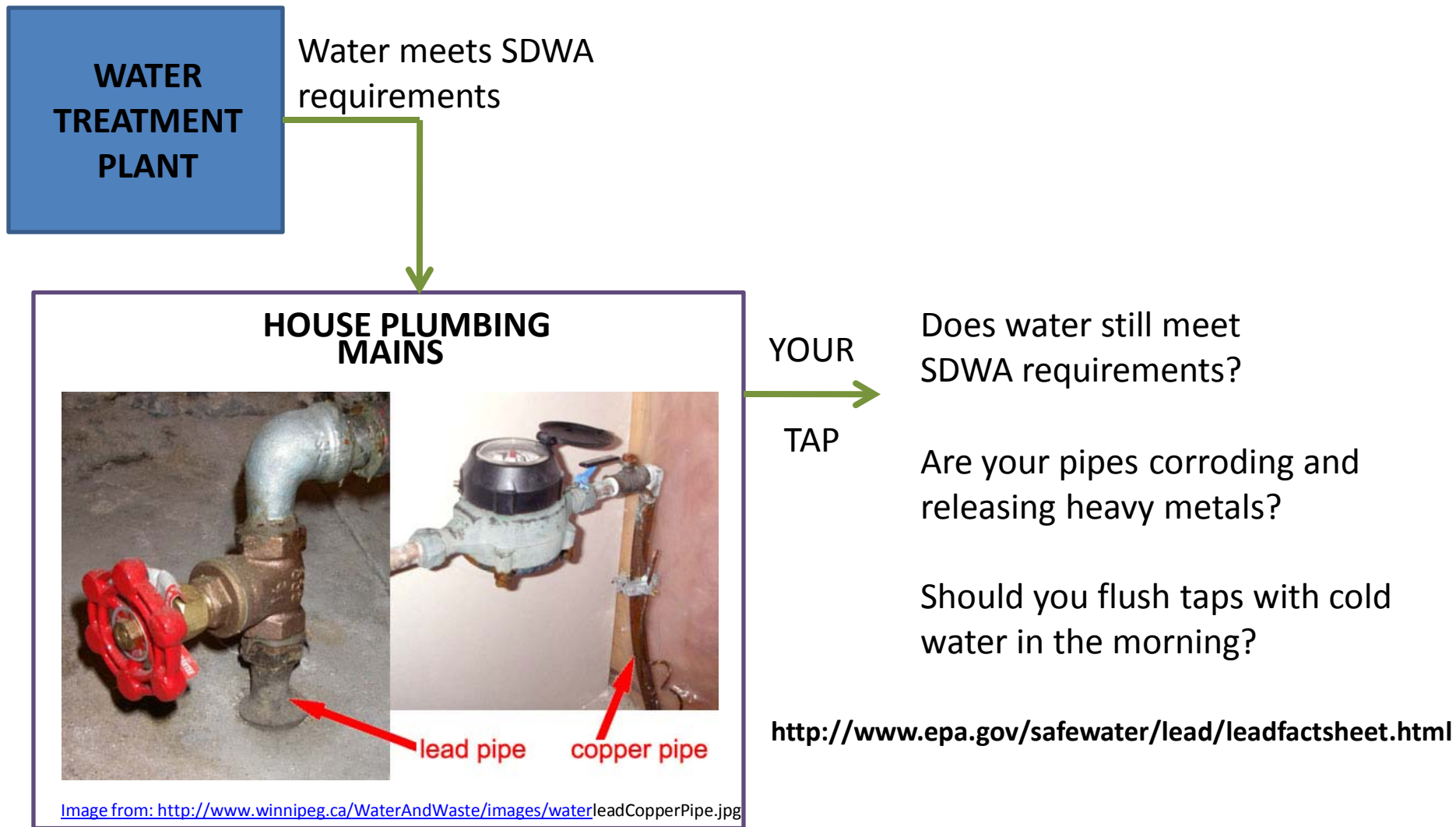
Howard F. Curren Wastewater Treatment Plant: Cu data



Howard F. Curren Wastewater Treat Plant: Ni data



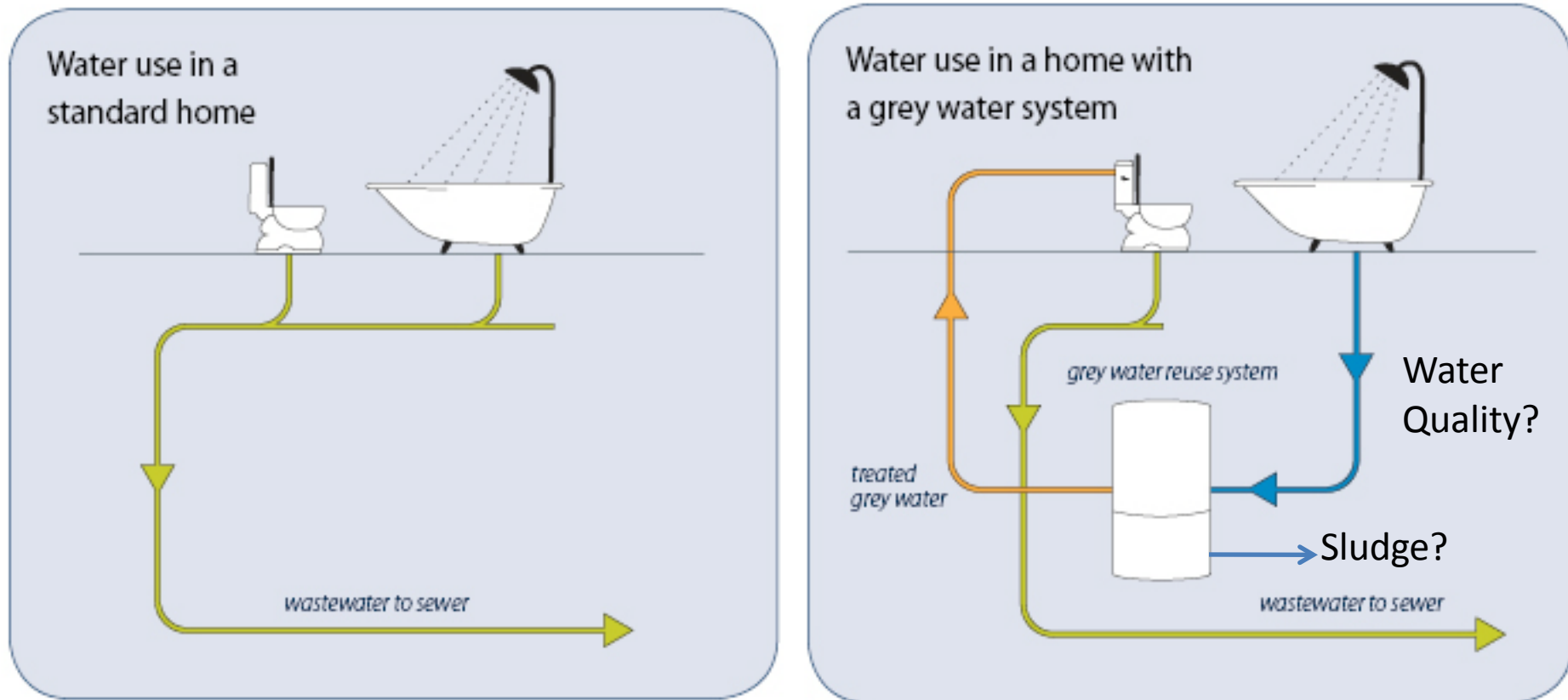
What about water out of your tap?



Changes in water quality as a function of time & location in the distribution system

Heavy metals in grey water

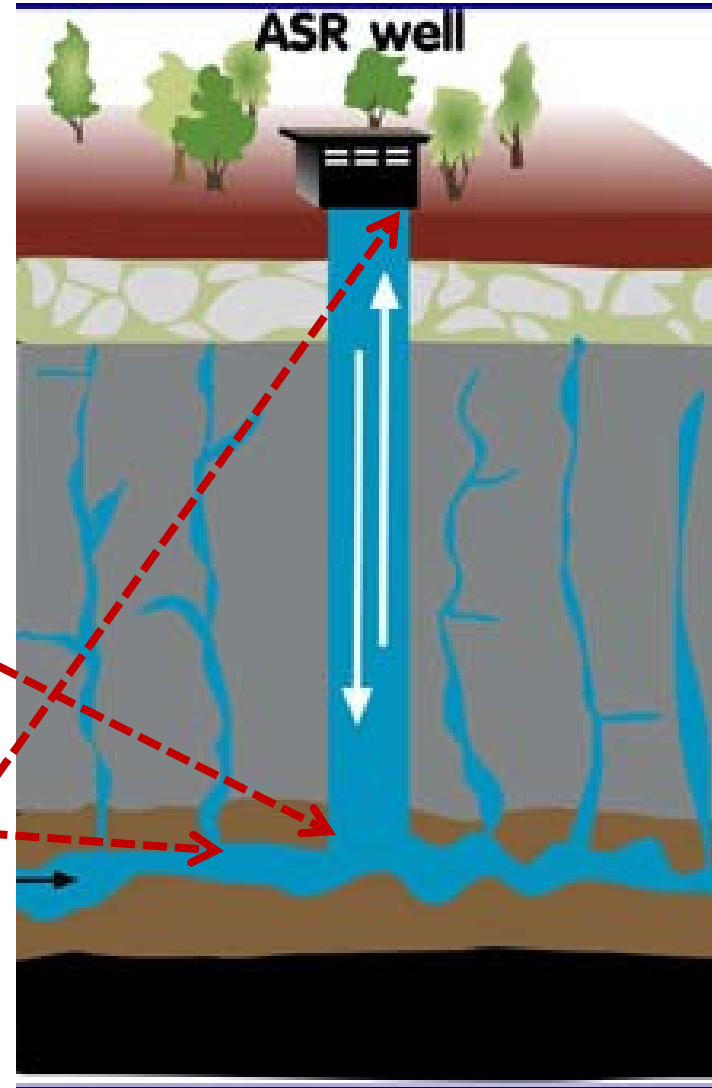
Potential sources in the house: plumbing materials, cutlery, jewelry, coins, home maintenance products, arts and craft products, dental fillings.



Changes in water quality as a function of household & treatment option

Aquifer Storage & Recovery (ASR)

- Storage of water in an aquifer to be used later
- Proposed injection of wastewater effluent
 - Water quality of injected water different from groundwater
 - Changes in groundwater concentrations possible
 - Dissolution/precipitation reactions
 - Possible increase in heavy metal concentrations
 - Additional treatment needed to remove heavy metals once pumped



Conclusions

- Heavy metals refers more to metals (transition, alkali earth, semi metals) in the periodic table that cause some toxic effect and are regulated.
- Heavy metals are used in many products & they never degrade.
- Worldwide bans on mercury and lead use can reduce our exposure to these metals.
- Various guidelines exist for aqueous heavy metal concentrations depending on the intended use of the water.
- Concentrations of heavy metals in wastewater effluent depend on treatment employed and at the Howard F. Curren wastewater treatment plant they are usually on the order of or below drinking water standard concentrations.
- Grey water systems can potentially expose us to higher concentrations of heavy metals than traditional sewer systems.
- We may be exposed to high concentrations of heavy metals through changes in water quality in the distribution system, especially due to corrosion.
- Aquifer Storage and Recovery processes can alter groundwater concentrations of heavy metals. This will require additional treatment when we wish to use it, but for most cases the technology exists.

Thank You.

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